SYNTHESIS OF 1-AZASPIRO[5.5] UNDEC-7-EN-2-ONE

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The 1-azaspiro[4 5]decane (1) and 1-azaspiro[5 5]undecane (2) ring systems are important structural features of a variety of natural alkaloids 1 In addition, synthetic

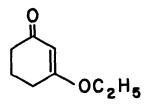


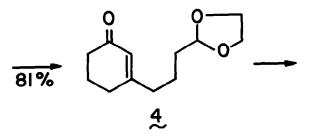
compounds with these ring systems have been reported to display a broad spectrum of pharmacological activities ² Although a fairly large number of synthetic routes exist for the synthesis of 1-azaspiranes functionalized in the azacyclic ring, ³ synthetic approaches which allow for versatile incorporation of functionality in the carbocyclic ring are conspicuously absent. In this letter, we report a potentially general route to such systems which regiospecifically incorporates unsaturation in the carbocyclic ring

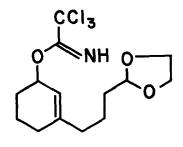
In particular we wish to describe a short route to the unsaturated spirolactam $\underline{3}$, which is based on our recently reported⁴ method for the 1, 3-transposition of oxygen and nitrogen functionalities Spirolactam $\underline{3}$ contains synthetic handles for further elaboration of functionality at carbons 2, 7, and 8, and thus is a potential precursor to the ring system of the anticholinergic, histrionicotoxin alkaloids¹

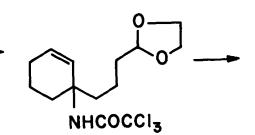
Grignard addition of readily available⁵ 1-ethylenedioxy-4-chlorobutane to 3-ethoxy-2cyclohexenone afforded, after hydrolysis at pH 1 5, enone 4⁶ in 81% yield, b.p 135-140 (.01 torr). Reduction of $\underline{4}$ with LiAlH₄ (2 equiv) in THF at -78° yielded quantitatively the corresponding Treatment of this alcohol with sodium hydride (15 equiv) in ether, followed by alcohol addition at 0° of the resulting alcohol-alkoxide solution to an ethereal solution of trichloroacetonitrile (1 1 equiv) afforded the allylic trichloroacetimidate 5, ⁶ which was not purified, but directly thermally rearranged by heating at 69° in hexane for 4-5 days After separation of the insoluble by-product, trichloroacetamide (45-55% yield), the rearranged amide 6 crystallized directly from the concentrated hexane solution; 20% overall yield from 4; mp 90 5-91 5°; ir (KBr) 3200, 1701, and 1511 cm⁻¹, pmr (CCl₄) τ 3 72 (1<u>H</u>, broad s, N<u>H</u>) and 4 11 (2<u>H</u>, broad s, = CH) No trace of the allylic isomer of 6 could be found upon careful examination of the crude reaction mixture by pmr The major products formed, in addition to 6, are a mixture of dienes resulting from the loss of trichloroacetamide from 5 Oxalic acid catalyzed hydrolysis of $\underline{6}$ in acetone-water (4 1) followed directly by oxidation with AgO⁷ afforded acid $\frac{7^6}{6}$ (mp 110-111 5°) in essentially quantitative yield. Treatment of 7 at 25° with 3M NaOH for 2 hr, followed by removal of the water and esterification $(H_2SO_4$ in methanol) gave the corresponding amino ester which spontaneously cyclized when basified to afford spirolactam 3;^{6, 8} 69% overall yield from 6; mp 117-118°; ir (KBr) 3190, 3050, 1663 cm⁻¹; pmr (CCl₄) τ 2 41 (1H, broad s, NH), 4 13-4 67 (2H, m, = CH)

The yield of amide $\underline{6}$ from the thermal rearrangement of imidate $\underline{5}$ is disappointingly low, and in fact is the lowest we have observed for a secondary allylic trichloroacetimidate ⁴ However, the high yields of the other steps and the fact that only two intermediates, $\underline{4}$ and $\underline{6}$, need be isolated and purified, combine to make this scheme nevertheless efficient We anticipate that the large amount of elimination which accompanies the [3 3]-sigmatropic rearrangement of trichloroacetimidate $\underline{5}$ may be suppressed in imidates bearing less electron withdrawing acyl substituents



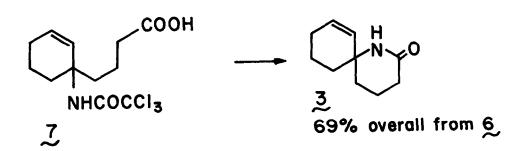






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6 20% overall from 4



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- 8 Catalytic hydrogenation afforded 1-azaspiro[5 5]undecan-2-one, which was identical with an authentic sample³ kindly provided by Professor R K Hill